Nitrogen Transformations in Soils Previously Amended with Sewage Sludge

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ABSTRACT

This short-term (10-d) incubation experiment established the rates of nitrogen (N) transformations occurring in sludge-amended and nonamended soil. Utilizing a nitrification block (C2H2) with (15NH₄)₂SO₄, first-order rate constants were calculated for N immobilization, ammonification, nitrification, and denitrification. These rate constants were compared to values obtained after a long-term (87-wk) incubation performed on soils sampled from the same field plots. The short-term rates of ammonification were still higher than the controls 4 yr after the last sludge addition. Sludge applications over an 8-yr period (180 Mg ha-1 yr-1) reduced soil nitrification potential compared to the controls when spiked with 15N. Denitrification did not cause a significant loss of N during either a shortor long-term incubation period. The microbial biomass in the sludgeamended soil contained more N, which resulted in a microbial C/N ratio of approximately 4:1 vs. 5:1 for the controls, Initial (shortterm) N immobilization rate constants were 0.43 for the sludgeamended and 0.35 for the nonamended soil.

Soil Microorganisms function as organic "microprocessors" of the terrestrial ecosystem by facilitating the nutrient flow and decomposition of organic residues. Currently, there is a need to improve the integration of data concerning these microbes and the soil N processes they control.

The hub of N transformation in soil is the NH[‡] pool, for it lies at the crossroads of three major N processes: nitrification, ammonification and immobilization. Soil N mineralization rate, a measurement of gross ammonification minus immobilization, has traditionally been determined by the accumulation of inorganic N, mainly NO₅. However, acetylene (C₂H₂), which inhibits nitrifying bacteria such as Nitrosomonas spp., can be employed to determine the ammonification rate by measuring the accumulation of ammonium. Acetylene also allows for the determination of N uptake by the microbial biomass in the absence of the competing nitrification reaction. With

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the use of ¹⁵N in conjunction with the chloroform fumigation incubation method (CFIM), the flux of N into soil microbial biomass can be measured (Jenkinson and Powlson, 1976; Voroney and Paul, 1984). An estimate of the denitrification rate can also be obtained by measuring the increase in N₂O over a short-time period when C₂H₂ is used to inhibit nitrous oxide reductase (Smith et al., 1978).

The purpose of this work was to evaluate the effects of sludge on long-term N processes utilizing a short-term N transformation study. The objectives were to: (i) measure microbial-N and determine the short-term flux of N into the soil microbial biomass; (ii) estimate soil ammonification and nitrification rates; (iii) determine if N mineralization (net ammonification) constant (k) is indeed constant; and (iv) determine the significance of denitrification during aerobic incubations.

MATERIALS AND METHODS Field Study

The 2.4- by 3.0-m plots are located on the Oxford Tract at the Univ. of California, Berkeley. The Tierra loam soil had an original CEC of 20.1 cmol kg⁻¹ and a pH of 5.4. A municipal sludge (Oakland) was incorporated into triplicate plots annually (180 Mg ha⁻¹ yr⁻¹) for 8 yr with no addition in the subsequent 4 yr. A crop of barley was grown on the site each of the 12 yr of the study. The sludge was anaerobically digested for 20 d then vacuum-filtered. When applied to the field, the sludge was a wetcake slurry that contained 25% solids (Williams et al., 1984).

Short-Term Laboratory Incubation Study

Surface soils (0–15 cm) from two control (check) and two sludge-amended plots were collected 4.5 yr after the last sludge application. The field moist soils were sieved (<4 mm) and bulked into either check or sludge-treated samples. These soil samples (20-g oven-dry weight) were mixed with 20 g of Ottawa sand (0.59–0.42 mm) to aid filtration. Labeled (15NH₄)₂SO₄ (70.5 atom % 15N excess) solution was then applied with a syringe at a rate of 20.8 mg (14N + 15N)-N kg⁻¹ soil to each soil/sand sample. Water was added to bring the soil-sand mixture up to -100 kPa water potential (60% water-holding capacity). The soil-sand mixtures were placed into 236-mL Mason jars fitted with a gas sampling septum.

Half of the jars were injected with C₂H₂ (30 cm³) to inhibit both nitrification and nitrous oxide reductase. Measurements of N₂O were taken 1, 3, 5, and 24 h after injection of C₂H₂ (5 mol m⁻³) on a Varian Model 3700 gas chromatograph (Varian, Palo Alto, CA) equipped with a ⁶³Ni electron capture detector (ECD) (Smith et al., 1978; Strauss, 1983).

Triplicated soil samples, with and without the nitrification block (15 cm 3 C $_2$ H $_2$), were then incubated for an additional 2 or 9 d at 25 °C. The N pool sizes were measured at 0, 3 and 10 d in an attempt to frame the most important changes in N uptake and nitrification. At the end of the incubation, the soils were shaken for 30 min with 75 mL of 0.5 M KCl and extracted to determine inorganic-N content. The soils were rinsed with 25 mL of deionized water, in an attempt to remove excess salt, and were then vacuum-filtered back to -100 kPa.

Half of the samples were fumigated with CHCl₃ for 24 h and allowed to incubate at 25 °C for an additional 10 d to determine microbial biomass C from the evolved CO₂ according to Jenkinson and Powlson (1976). The CO₂ evolved from the nonfumigated soils was not subtracted from the fumigated samples because an appropriate control for biomass C has yet to be determined (Voroney and Paul, 1984). To determine biomass N, the soils were extracted 10 d after fumigation with 75 mL of 2.0 M KCl and measured for inorganic N (Voroney and Paul, 1984). Also at this time the microbial N extract was measured for ¹⁵N content to determine the N immobilization rate.

Soil extracts were also performed on soils immediately after the addition of (15NH₄)₂SO₄ to determine the recovery efficiency of the labeled N and the standing pool sizes of inorganic N. The efficiency of recovery of the added 15NH₄ was found to be 91.3% in the sludge-amended soil.

Total NH₄-N and NO₃-N in the KCl extracts were determined by steam distillation followed by autotitration on a Fisher Model 381 (Fisher Scientific, Pittsburg, PA) using 0.0128 M H₂SO₄. The following sequence was followed to avoid cross contamination of the isotope ratio measurements during steam distillation:

- 25 mL of 5% acetic acid was distilled then discarded;
 25 mL of 90% ethyl alcohol was distilled then dis-
- carded;
 3 80 mL of 0.5 M KCl extract was distilled with MgO:
- 3. 80 mL of 0.5 M KCl extract was distilled with MgO;
- 40 mL of distillate was collected in a glass beaker with 5 mL of 2% H₃BO₃;
- 5. step 1 and 2 were repeated;
- Devarda's alloy was added to the 0.5 M KCl extract in step 3 distilled again and the 40 mL of distillate was collected in 5 mL of 2% H₃BO₃;
- the 2.0 M KCl extracts were analyzed similarly, except that MgO and Devarda's alloy were added together and distilled only once to determine combined NH₄-N and NO₃-N content;
- the samples were acidified with 2 mL of 0.04 M H₂SO₄ and dried in a 60 °C oven that was flushed with acidscrubbed air; and
- the samples were transferred to test tubes with successive washings of methanol and water. The samples were again dried then sent to Los Alamos for ¹⁵N mass spectrometric analysis (Hauck, 1982).

RESULTS AND DISCUSSION

Biomass Nitrogen

More N was found in the sludge-amended biomass than in the check biomass, reflecting the higher availability of inorganic N in the sludge-amended soil (Fig. 1). The C/N ratio of the microbial biomass was lower in the sludge-amended soils, averaging 3.9 vs. 4.8 for the checks. These biomass C/N ratios are lower than

those reported by Anderson and Domsch (1980) but are consistent with the ratios of Voroney and Paul (1984). The low microbial C/N ratios suggest that the immobilized N conserved in the microbial biomass could allow for inorganic N to be released slowly over time as the microbial C/N ratio increases and the general microbial biomass decreases.

After a long-term incubation (initiated 87 wk earlier), the C/N ratio of the microbial biomass of the check soil rose to 6.2 and in sludge-amended soil to 5.7 (Boyle, 1986). A decline in sludge microbial biomass C from 409 to 260 mg kg⁻¹ was also observed in the field during this time (Boyle and Paul, 1989).

The short-term immobilization of ¹⁵NH₄ and ¹⁴NH₄ into the microbial biomass of both soils is depicted in Fig. 2. The immobilization of N was greater in the check soil in the presence of C₂H₂ than when nitrification was allowed to proceed, which is consistent with results of Nishio et al. (1985). Nitrification seemed to be competing with N immobilization in the check soil, and in the sludge-amended soil after 3 d.

Inorganic Nitrogen

The distribution of soil N among the biomass and inorganic pools is depicted in Fig. 3 and 4. In both soil treatments, the size of the NH₄ pool increased from day 3 to day 10 with the nitrification block. The increase can be attributed to the mineralization of or-

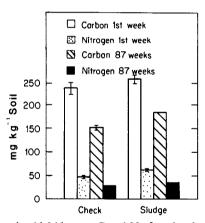


Fig. 1. The microbial biomass C and N after the short-term incubation as compared to a long-term incubation (Boyle and Paul, 1989). (Error bars represent the standard error of the mean).

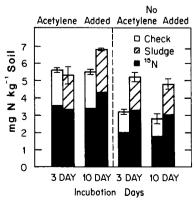


Fig. 2. The short-term uptake of N (14NH₄⁺ + 15NH₄⁺) into the soil microbial biomass.

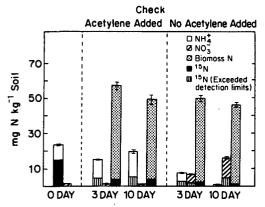


Fig. 3. The N pool sizes of the check soil with and without acetylene at 3 and 10 d.

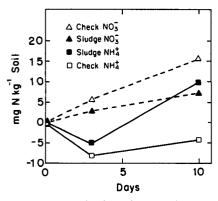


Fig. 5. The short-term production of NH_4^* (with C_2H_2) and NO_3 (without C_2H_2).

ganic N to NH₄. Without the C₂H₂, the NH₄ pool size decreased significantly in both soils between 0 and 10 d.

The NO_3 analysis of the 0.5 M KCl solution indicates that the C₂H₂ block was more effective for the check soil, in which there was little change in NO₃ present after 10 d, than for the sludge-treated soil in which the NO₃ concentration was as high with the C₂H₂ block as without the block (Fig. 3 and 4). The sludge-treated soil had initially (day 0) higher concentration of NO₃ than the check soil which may have obscured the response of the block. However, the effectiveness of the nitrification block was confirmed by the paucity of ¹⁵NO₃ found in the presence of C₂H₂. Reliable 15NO3 results could not be obtained from the check soil, which contained a small unlabeled NO3 pool, because the high ¹⁵N enrichment of this pool approached or exceeded the detection limit of the mass spectrometer (approx. 30%)

The term "mineralization" includes both ammonification and nitrification processes, however in most soils the limiting step is the conversion of organic N to NH₄. The 10-d increase in net production by both these processes is depicted in Fig. 5 (accumulation—initial concentration). The NH₄ production in soils treated with C₂H₂ is represented by the solid lines. Both soils displayed a net negative production of NH₄ at day 3 which demonstrated the initial dominant effect of immobilization.

The dashed lines in Fig. 5 represent "potential" nitrification because both soils were spiked with 20.8 mg

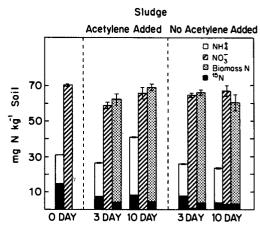


Fig. 4. The N pool sizes of the sludge-amended soil with and without acetylene at 3 and 10 d.

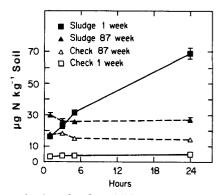


Fig. 6. The production of N₂O-N (with C₂H₂) at day 1 and after 87 wk of incubation.

NH₄⁻N kg⁻¹ soil. The reduction in nitrification potential in the sludge-amended soil could be due to the lower pH (pH 4.9 vs. 5.6 for the check soil), greater metal content (422 mg Zn kg⁻¹ soil vs. 114 mg kg⁻¹ for the check) or a combination of the two.

It has been suggested that denitrification is a major cause for the loss of N in some aerobic incubations (Ryan et al., 1973). Lindemann and Cardenas (1984) reported up to 65% of mineralized-N lost in sludgetreated soils through denitrification. These authors attributed the nonlinear increase of NO₃ production with increased sludge additions to be due in part to denitrification. The data presented here (Fig. 6) does not support this route as a significant loss of N during aerobic incubation in these soils. Denitrification as determined by N₂O accumulation with C₂H₂ block was essentially zero for the check soil, while the sludgeamended soil produced 68 µg N kg⁻¹ in the first 24 h of the short-term incubation (Fig. 6). Because denitrification was measured by the amount of N₂O produced from the NO₃ soil pool (each point represents an average of eight measurements), the low initial NO₃ pool size of the short-term check soil (0.5 mg N kg⁻¹) could have caused the low rates. However, for the long-term 87-wk incubation, both the check and sludge-soil were naturally enriched with higher levels of NO₃ (approximately 30 and 70 mg N kg⁻¹, respectively), but neither soil produced an increase in N₂O levels in the 24-h period. These results suggest that the low production of N₂O was due to factors other

Table 1. The comparison of first-order rate constants for four soil N transformations.

	Short-term (10 d)		Long-term (87 wk)	
	Check	Sludge	Check	Sludge
	(week-1)			
Immobilization	0.64	0.42	0.26	0.24
without C2H2	0.35	0.43	ND‡	ND
Ammonification			-	
net	†	0.06	0.01	0.02
gross	0.03	0.10	ND	ND
Nitrification				
potential	0.69	0.18	ND	ND
Denitrification	0.00	0.04	0.00	0.00

^{† =} between 0 and 10 d the net ammonification rate remain less than zero, but between 3 and 10 d the rate was 0.19 (0.12 for the sludge soil).

‡ ND = not determined.

than low NO₃ content, possibly the scarcity of anaerobic sites or available C.

Nitrogen Transformation Rates

Because first-order kinetics have been successfully used to characterize soil N processes (Stanford and Smith, 1972; Myrold and Tiedje, 1986), first-order rate constants (k) for N transformations during the short incubation period (10-d) and the constants obtained from the long-term (87 wk) incubation sampled a year and a half earlier from the same field plots (Boyle and Paul, 1989) were calculated and are presented in Table 1.

Nitrogen immobilization rates were calculated from 0- to 3-d incorporation of $^{15}NH_{4}^{+}$ into the microbial biomass (Voroney and Paul, 1984). The k value was found to be slightly greater for the check soil than the sludge-soil if nitrification was blocked by C_2H_2 (Table 1). The lower k value for the check soil in the absence of C_2H_2 corresponds to the lower N content in the microbial biomass (Fig. 1). After 87 wk, the value of k decreased to about 0.25 wk⁻¹ for both soils (Table 1).

The check soil exhibited a lower NH₄ production than the sludge soil (Fig. 5), in fact after 10 d this soil did not attain its initial (0-d) NH₄ pool size. For both check and sludge-amended soil, the net and gross ammonification rate constants were determined between day 0 and day 10. The gross ammonification rate was determined by adding the immobilization uptake of N to the net production of NH₄. To determine shortterm ammonification rate, the initial pool size has to be indirectly measured because it represents the portion of the organic N that is potentially mineralizable. An estimate of this short-term potentially mineralizable N pool size was calculated by fractionating the long-term (87-wk) incubation data into two pools (Boyle and Paul, 1989). The short-term (10-d) net ammonification rate constants were similar to the k values obtained from curve splitting the long-term data into an 11-wk pool (Boyle and Paul, 1989). These shortterm ammonification k values are also similar to the mineralization values reported by Myrold and Tiedje (1986). After 87-wk, the ammonification constants were significantly lower (Table 1) indicating that organic N mineralization should be characterized as having more than one first-order rate constant.

The nitrification potential rate constant was greater for the check soil than for the sludge-amended soil (Table 1). The check rate of nitrification was determined by 3-d (¹⁴N + ¹⁵N) NO₃ accumulation while the sludge nitrification was calculated by the less variable ¹⁵NO₃ data.

The term "production rate" represents the initial pool size minus the change in pool size over the 3-d incubation due to a particular N transformation [production rate = $N_{\text{(initial)}} (1 - e^{-kt})$]. For N immobilization, the production rate indicates the amount of decay of the NH₄ reservoir due to microbial accumulation. Microbial immobilization of NH⁺ was higher in the check soil than the sludge-amended soil only if nitrification was blocked. The production rate of NH₄ from organic N (with C₂H₂) was greater in the sludge-amended than the check soil which did not display a positive rate between 0 and 10 d. The potential for the production of NO₃ from the spiked NH₄ pool was lower in the sludge-treated soil than the check. Although it was noted that overall production of NO₃ is greater in these sludge-amended soils than the checks (Boyle and Paul, 1989), this is assumed to be due to the greater production of NH₄ from a larger organic N fraction, and not due to a greater rate in the nonlimiting nitrification step. The rate of denitrification and the production of N₂O production was initially greater in the sludge-amended soil, but this would represent less than 0.5 mg N kg⁻¹ lost from the soil per week, which is minimal compared to NO3 production.

SUMMARY

Previous applications of sludge (4 yr earlier) increased the short-term (10-d) soil ammonification rate.

The rate constant for ammonification (N mineralization) decreased between 10 d and 87 wk of incubation, indicating a need for more than one mineralization constant.

Eight years of sludge application (a total of 1440 Mg sludge ha⁻¹) decreased soil nitrification potential.

Denitrification was not found to be a significant loss of inorganic N from these soils during either a short or long-term incubation.

The initial short-term N immobilization rate constant was slightly higher (without C₂H₂) in the sludge-amended than the check soil.

The sludge-amended soil had a lower biomass C/N ratio than the check soil. Excess storage of N in microbial cells in the sludge-amended soil could provide a means of retaining N in the elastic labile fraction over an extended period of time (at least 4 yr).

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